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# General Physics II

## Electrostatic: Principles & Applications



### Lecture (1): Coulomb's Law

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Charles Coulomb  
1736 - 1806

## Coulomb's Law

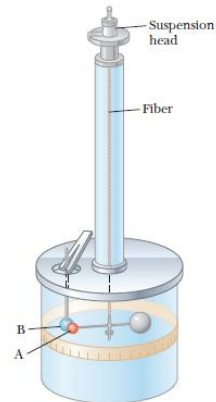
In 1785, Coulomb established the fundamental law of *electric force* between two stationary, charged particles. Experiments show that an electric force has the following properties:

(1) The force is *inversely proportional* to the square of separation,  $r^2$ , between the two charged particles.

$$F \propto \frac{1}{r^2}$$

(2) The force is *proportional* to the product of charge  $q_1$  and the charge  $q_2$  on the particles.

$$F \propto q_1 q_2$$



(3) The force is *attractive* if the charges are of opposite sign and *repulsive* if the charges have the same sign.

We can conclude that

$$F \propto \frac{q_1 q_2}{r^2} \longrightarrow \therefore F = K \frac{q_1 q_2}{r^2}$$

where  $K$  is the coulomb constant =  $9 \times 10^9 \text{ N.m}^2/\text{C}^2$ .

The above equation is called *Coulomb's law*, which is used to calculate the force between electric charges. In that equation  $F$  is measured in Newton (N),  $q$  is measured in unit of coulomb (C) and  $r$  in meter (m).

The constant  $K$  can be written as

$$K = \frac{1}{4\pi\epsilon_0}$$

where  $\epsilon_0$  is known as the *Permittivity constant of free space*.

$$\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2/\text{N.m}^2$$

$$K = \frac{1}{4\pi\epsilon_0} = \frac{1}{4\pi \times 8.85 \times 10^{-12}} = 9 \times 10^9 \text{ N.m}^2/\text{C}^2$$

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## Coulomb's Law

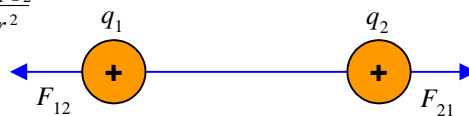
The electrostatic force of a charged particle exerts on another is **proportional** to the product of the charges and **inversely proportional** to the square of the distance between them.

$$\therefore F = K \frac{q_1 q_2}{r^2}$$

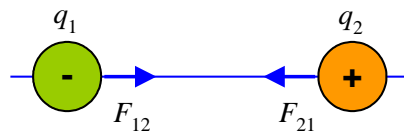
$$\vec{F}_{12} = -\vec{F}_{21}$$

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$$F_{12} = K \frac{q_1 q_2}{r^2} = F_{21}$$



**Repulsive force**



**Attractive force**

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## Example 1

Calculate the value of two equal charges if they repel one another with a force of 0.1N when situated 50cm apart in a vacuum.

### Solution

$$F = K \frac{q_1 q_2}{r^2}$$

$$0.1 = \frac{9 \times 10^9 \times q^2}{(0.5)^2}$$

$$q = 1.7 \times 10^{-6} \text{C} = 1.7 \mu\text{C}$$

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## Quiz 1

One charge of 2.0 C is 1.5m away from a -3.0 C charge. Determine the force they exert on each other.

## Quiz 2

Object A has a charge of  $+2\mu\text{C}$  and Object B has a charge of  $+6\mu\text{C}$ .  
Which statement is true?



- **A:**  $\mathbf{F}_{AB} = -3\mathbf{F}_{BA}$
- **B:**  $\mathbf{F}_{AB} = -\mathbf{F}_{BA}$
- **C:**  $\mathbf{3F}_{AB} = -\mathbf{F}_{BA}$
- **D:**  $\mathbf{F}_{AB} = 12\mathbf{F}_{BA}$

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